

# Some highlights and present work of the IFAR and possible interactions with ECO



Thomas P. Knepper

# Outline

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## **Introduction**

*Water Cycle*

*Past&Present*

*Selected Compounds*

## **Experimental Approach**

## **Results and Discussion**

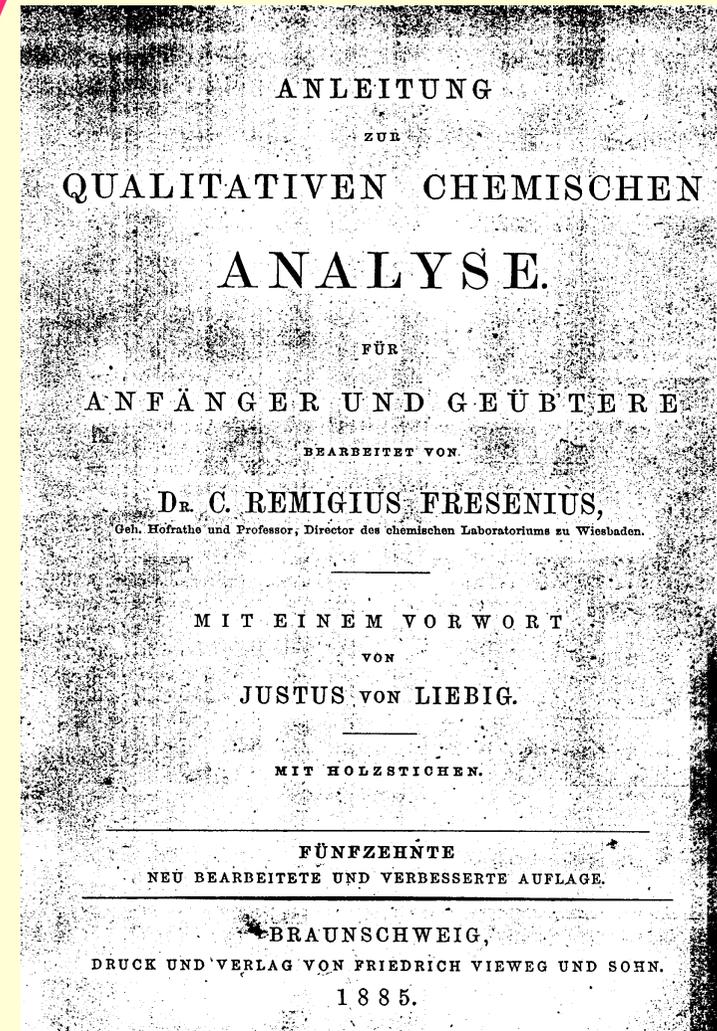
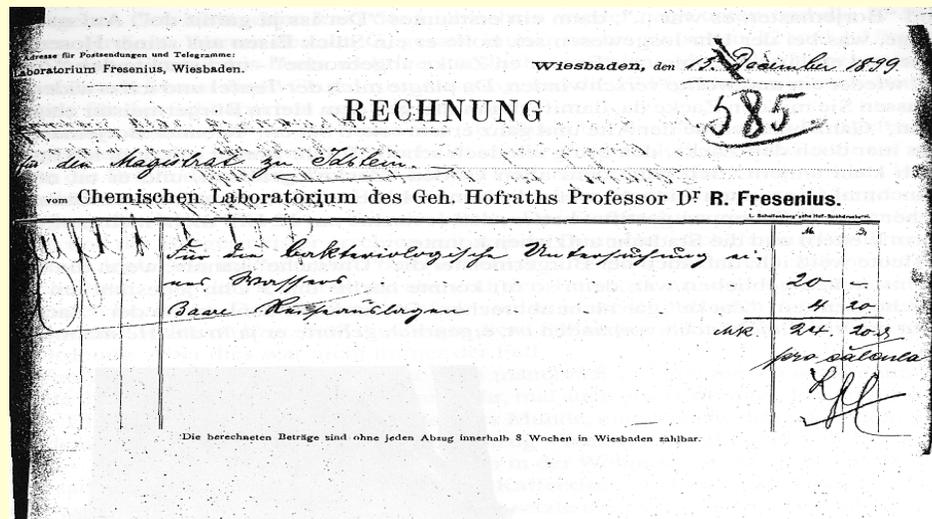
*Case Studies: Biodegradation, Photodegradation and Monitoring*

## **Conclusions & suggestions for future work within ECO**

# University of Applied Sciences Fresenius (HSF) Idstein, Germany

## History

1848 Foundation of „Chemischen Laboratorien Fresenius“



At present: approx. 3600 Students

# Department of Chemistry and Biology

## ***Vocational Education***

Chemical Technical Assistants

Biological Technical Assistants

## ***Continuing Education***

Chemical Technicians

Disinfector

## ***Study Courses***

Chemical Engineering  
(Degree)

### **Starting 2008/2009**

International Bachelor of  
Applied Chemistry/Bachelor of  
Science (BSc)

Bachelor Wirtschaftschemie

### **Starting 2009/2010**

Master of Bio- and  
Pharmaceutical Analysis

# Department of Chemistry and Biology

## Institute for Analytical Research (IFAR)



Equipment f.e.

GC-MS

LC-MS/MS Trap

Membrane bioreactor



### Aims

the essential products of teaching and learning, as well as practice and demand orientation, are guaranteed at the interface between teaching and practice

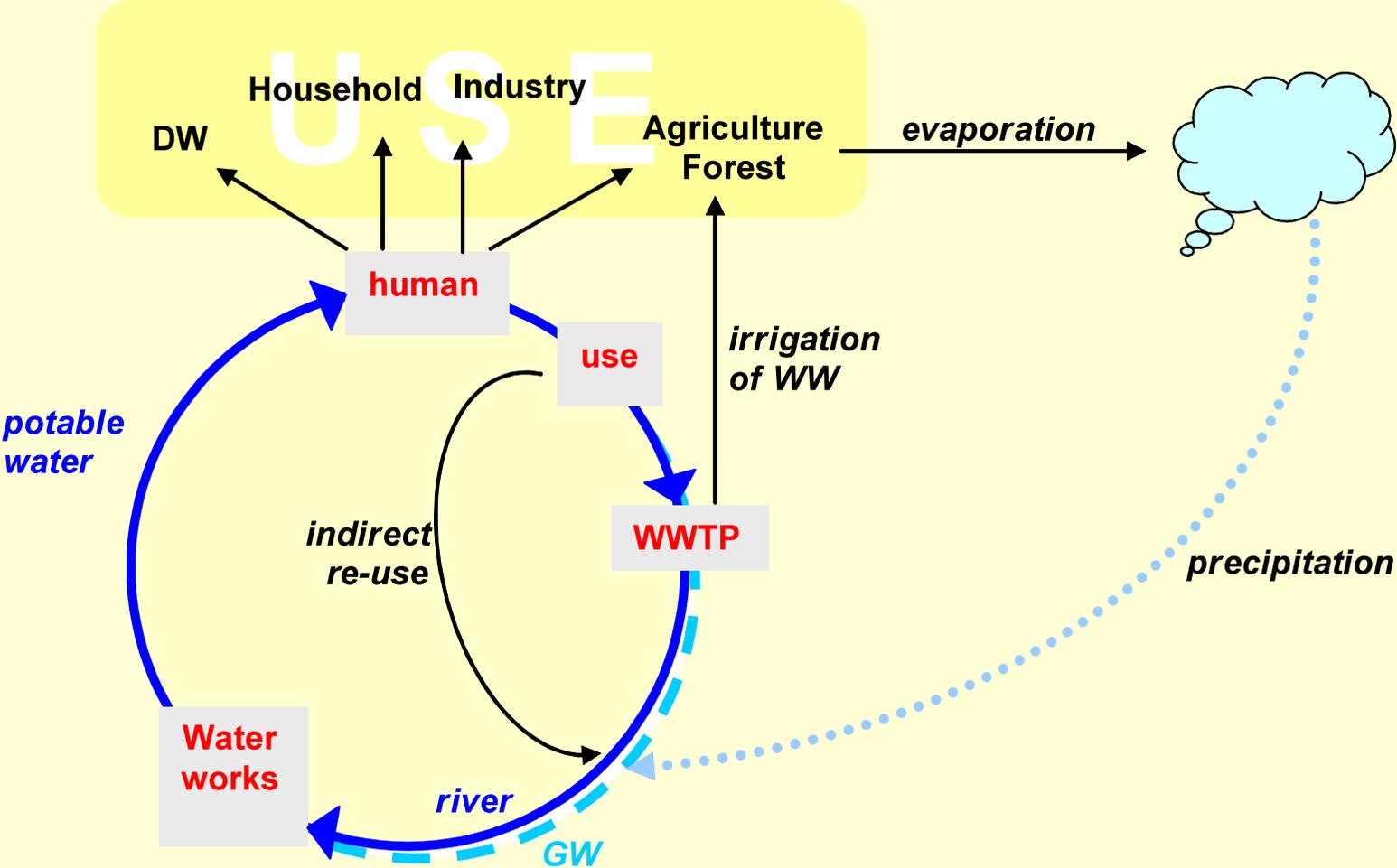
# **Innovative processes and practices for wastewater treatment and re-use in the Mediterranean region / INCO-CT-2006-517728**

## facts

- Supported by the European Commission
- Sixth Framework Programme
- Start date: 1. January 2007
- Duration: 42 months
- 13 participants



# Contaminants in the water cycle

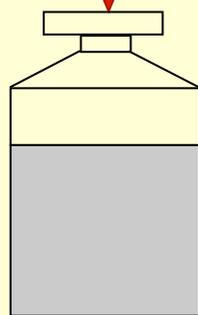


# Analytical Approach

Selected conaminants —  
— pharmaceuticals  
— pesticides  
— industrial contaminants  
— personal care products  
— surfactants

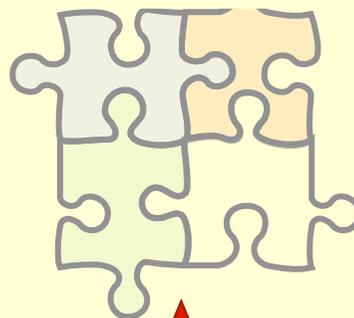
pharmaceuticals  
pesticides  
surfactants  
industrial contaminants  
personal care products

*What's known?  
What's not?*



**DEGRADATION  
TEST**

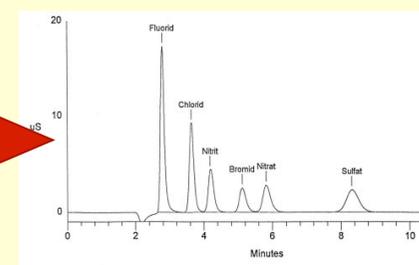
**IDENTIFICATION**



*mass  
spectrometry*

➤ degradate  
➤ parent  
compound

**ANALYTICAL METHOD**



*monitoring*

**fate in the  
environment**

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# MASS SPECTROMETRY

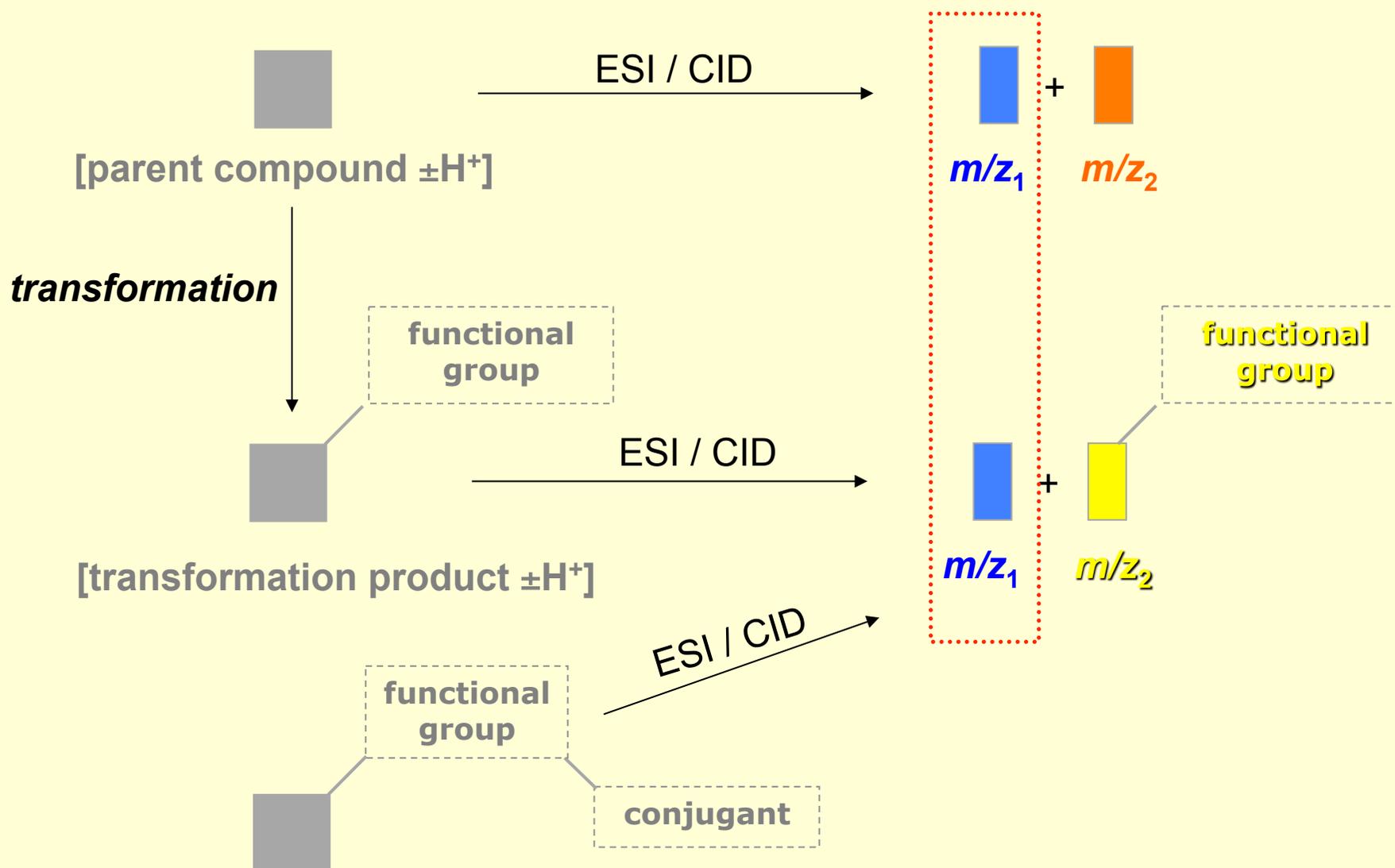
Structure  
Elucidation  
**(degradate)**

Sensitive  
Detection  
**(parent compound)**

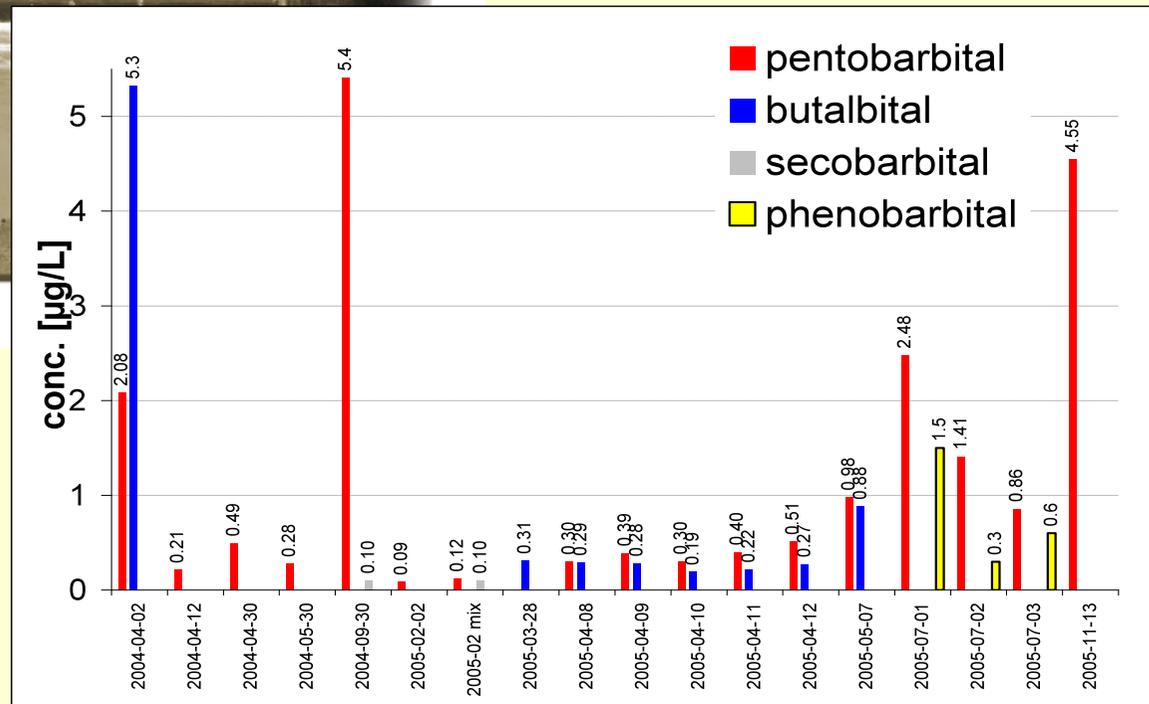
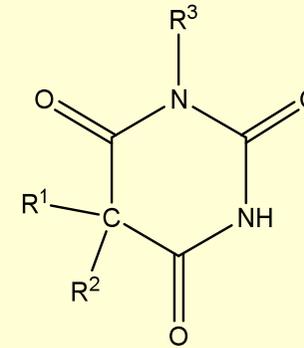
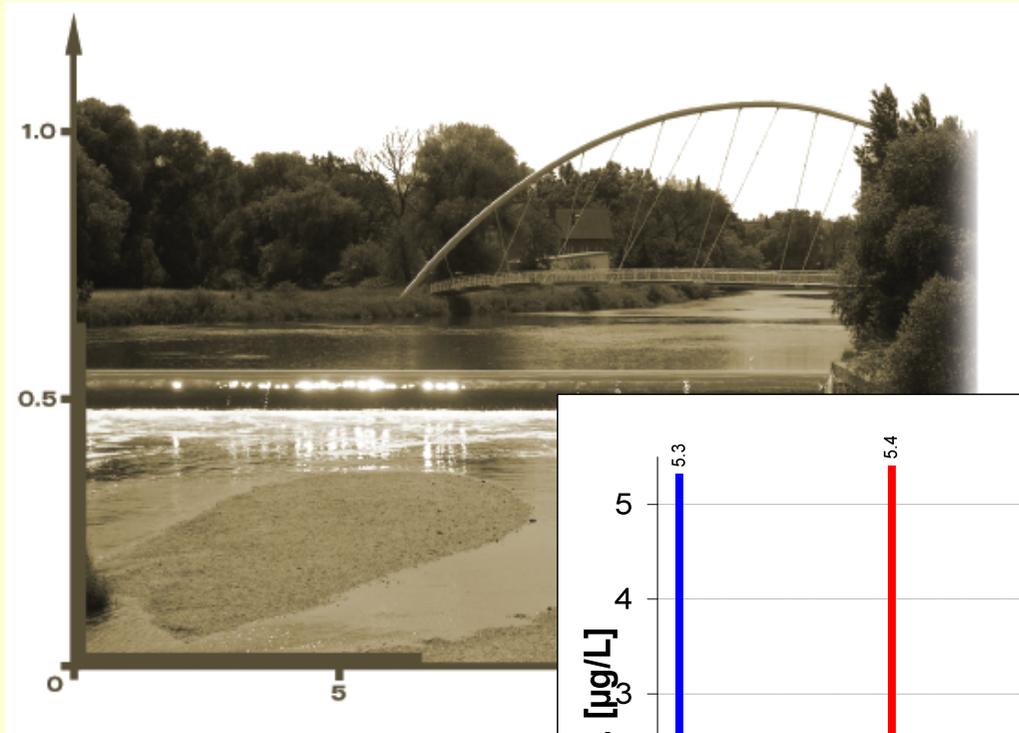
# Strategy to seek for transformation products

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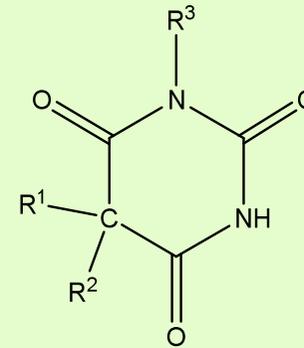
# Barbiturates



# Barbiturates

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- pharmaceuticals (sedative hypnotics)
- peak of consumption: mid-1960s
- $\log K_{OW} = 1.49 - 2.24$
- today only phenobarbital (antiepileptic) and
- thiopental (anaesthetics)



- pentobarbital and 5,5-diallylbarbituric acid present in groundwater (30 ng/L)<sup>1</sup>
- phenobarbital WWTP in effluent water (1000 ng/L)<sup>2</sup>

- **No biotic or abiotic biodegradability !**
- **Presence in groundwater , even after 40 years!**

<sup>1</sup> Eckel, Ross, Esensee; *Ground Water* 31, 1993

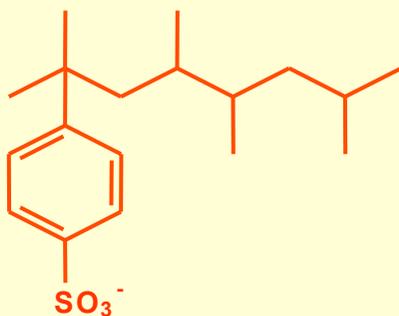
<sup>2</sup> Holm, Rügge, Bjerg, Christensen; *Environ. Sci. Technol.* 29, 1995

# LAS and ABS in surface and ground waters in the Philippines

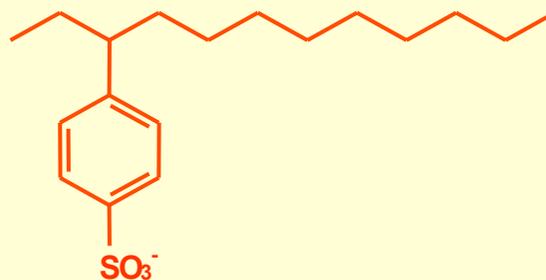


Surface Water sample	ABS (µg/L)	LAS (µg/L)	ABS/LAS ratio
Bucal Stream	3.5	25	0.14
San Pedro River	66	102	0.65
San Christobal River	9.2	8.4	1.1
Napindan Channel	1.0	2.2	0.45
Pasig River (Guadalupe)	4.1	8.0	0.51
Pasig River (Manila Bay)	22	27	0.81
Ground Water sample			
Spring – Drinking Water	3.2	0.35	9.1
Spring – Drinking Water-	7.2	2.6	2.8
Public Spring - Cabuyao	128	7.3	18
Spring – Cabuyao <sup>1)</sup>	253	4.9	52
Cabuyao Residential area	574	7.0	82
Cabuyao Highway, industrial	792	9.9	80

<sup>1)</sup> (intensive use for laundry washing between 1950 and 1970)



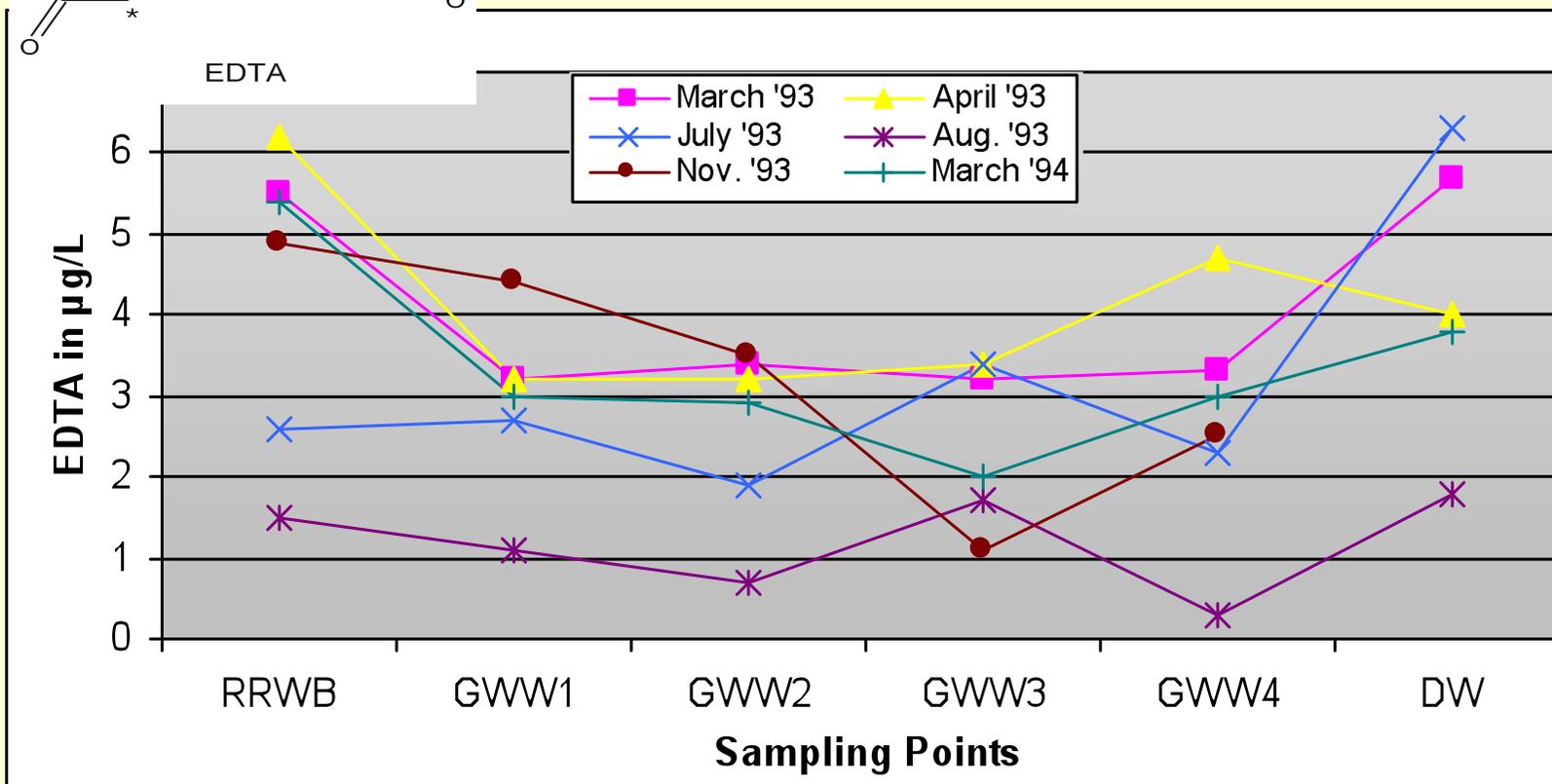
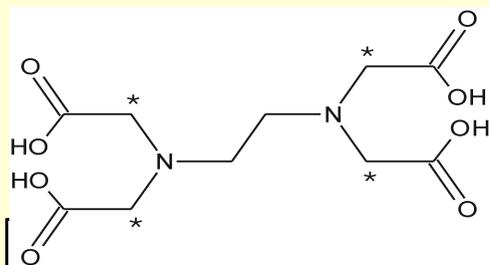
Branched alkylbenzene sulfonates  
(ABS)



Linear alkyl benzenesulfonates  
(LAS)

Eichhorn et al, Sci. Tot. Env. 2001;

# EDTA during ground water infiltration

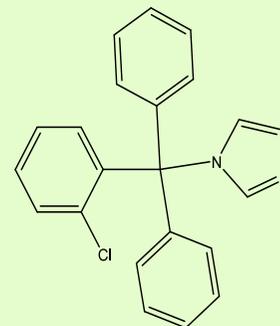


RRWB = River Rhine Water Basin; GWW1 = Groundwater Well (1 m);  
GWW2 (80 m); GWW3 (145 m); GWW4 (160 m); DW = Drinking Water

# Clotrimazole

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- pharmaceutical (antifungal agent)
- non polar ( $\log K_{OW} = 4.1$ )
- first authorised 1973
- consumed in Europe for more than thirty years with an annual quantity of approx. 20 t



- included on the OSPAR list of substances for priority action  
BUT: degradation test was performed at a concentration which exceeds water solubility<sup>1</sup>
- in surface water ( $\sim 5\text{ng/L}$ )<sup>2</sup> and WWTP influent and effluent (30 and 4 ng/L)<sup>3</sup>

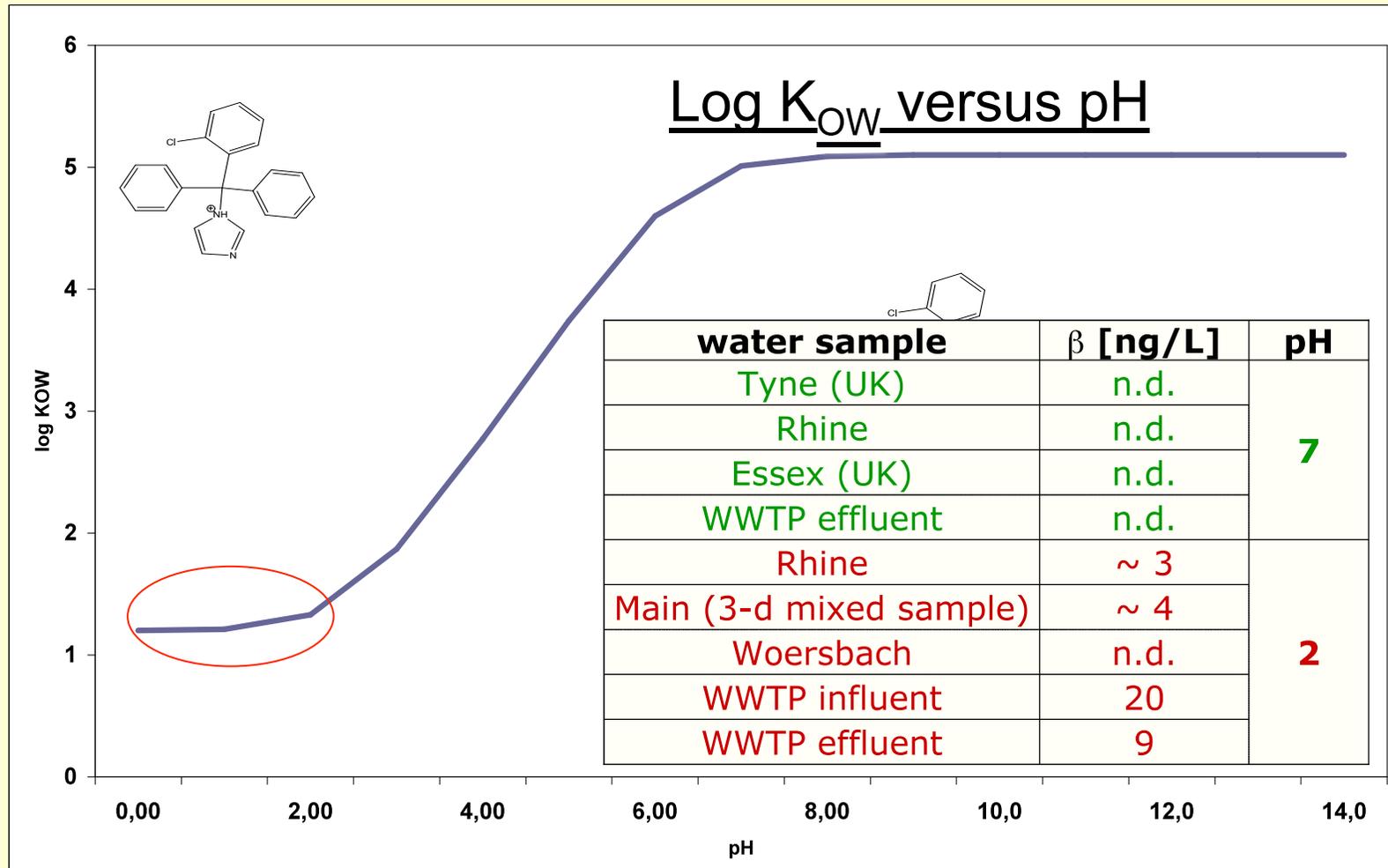
➤ **Can the P-criterion be verified?**

<sup>1</sup> OSPAR Commission; publication no 2005/199

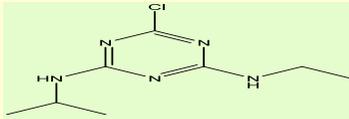
<sup>2</sup> Thomas, Hilton; *Mar Pollut Bull* 49, 2004

<sup>3</sup> Roberts; *Cefas contract report C2483*, 2006

# Monitoring of clotrimazole



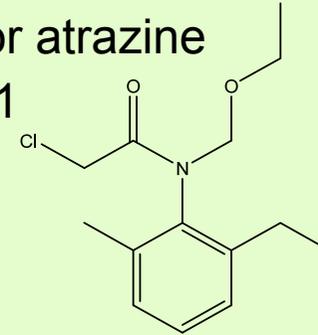
# Atrazine & Acetochlor



- herbicides

- substitute for atrazine

- $\log K_{OW}$  2.81



- 1950s start of application – 2004

- $\log K_{OW}$  2.63

- on the list of priority substances (WFD)

- known degradates: desethylatrazine, desisopropylatrazine, hydroxyatrazine<sup>1</sup>; oxanilic acid of acetochlor, ethansulfonic acid of acetochlor<sup>2</sup>
- atrazine and DEA in groundwater of the Bréville catchment<sup>3</sup> ( $0.2 \pm 0.7 \mu\text{g/L}$  and  $0.6 \pm 0.2 \mu\text{g/L}$ )

➤ **Does the soil in Bréville act like a reservoir for atrazine?**

<sup>1</sup> Qiao, Ma, Hummel; *Agric. Food Chem.* 44, 1996

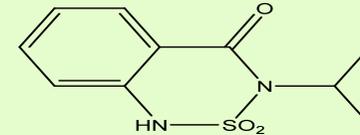
<sup>2</sup> Dictor, Baran, Gautier, Mouvet; *Chemosphere* 71, 2008

<sup>3</sup> PEGASE Final report BRGM/RP-52897-FR, 2004

# Bentazone

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- herbicide
- very polar ( $\log K_{OW} = -0.46$ )
- herbicidal activity first reported in 1968
- one of the most applied herbicides in the Ebro river delta



- 31  $\mu\text{g/L}$  during application time (mean value); Ebro 0.27  $\mu\text{g/L}$ <sup>1</sup>
- comparable high concentrations in river waters of the Tiber region (Italy)<sup>2</sup>
- not biodegradable in a fixed bed bioreactor<sup>3</sup>
- passes through membrane bioractor<sup>3</sup>
- present in ground- and surface water<sup>4</sup>

➤ **If not biodegradable → photodegradation?**

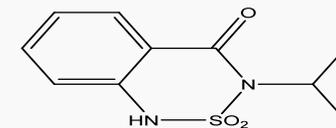
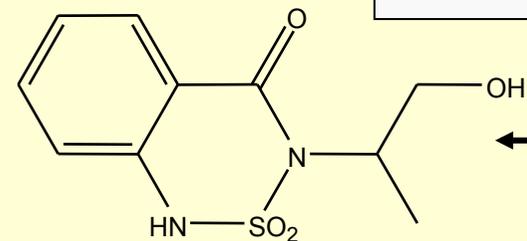
<sup>1</sup> Terrado, Kuster, Raldúa, Lopez de Alda, Barceló, Tauler; *Anal Bioanal Chem* 387, 2007

<sup>2</sup> Laganà, Bacaloni, De Leva, Faberi, Fago, Marino; *Anal Chim Acta* 462, 2002

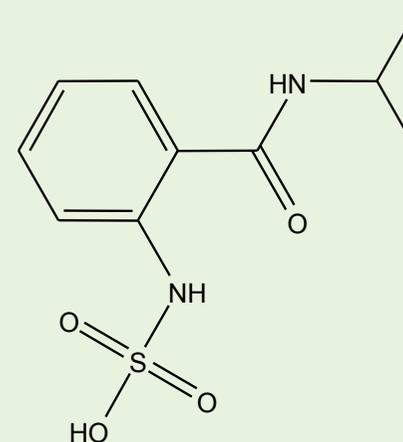
<sup>3</sup> González, Müller, Petrovic, Barceló, Knepper; *Environ Pollut* 144, 2006

<sup>4</sup> M. Isenbeck-Schröter, Auftreten von Pflanzenschutzmitteln  
in Oberflächengewässern und im Grundwasser

observed mass (m/z)	elemental composition actual mass	$\Delta$ mDa
255.0432 $\pm$ 0.0013	C <sub>10</sub> H <sub>11</sub> N <sub>2</sub> O <sub>4</sub> S m/z = 255.0440	- 0.8
197.0044	C <sub>7</sub> H <sub>5</sub> N <sub>2</sub> O <sub>3</sub> S m/z = 197.0021	2.3
191.0826	C <sub>10</sub> H <sub>11</sub> N <sub>2</sub> O <sub>2</sub> m/z = 191.0821	0.5
132.0320	C <sub>7</sub> H <sub>4</sub> N <sub>2</sub> O m/z = 132.0324	- 0.4



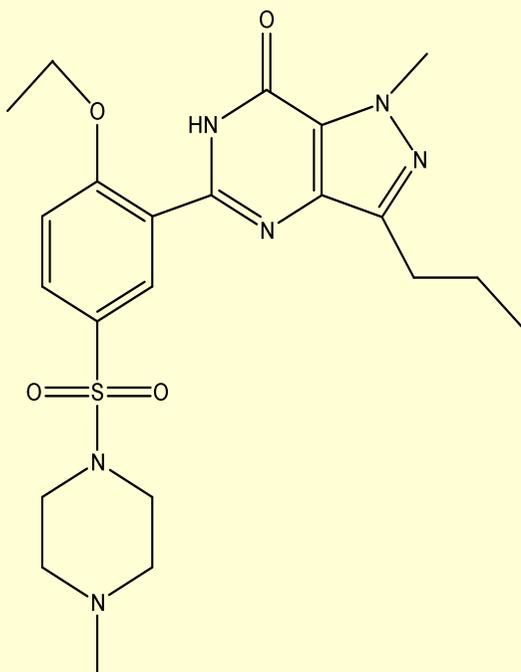
observed mass (m/z)	elemental composition actual mass	$\Delta$ mDa
257.0584 $\pm$ 0.0034	C <sub>10</sub> H <sub>13</sub> N <sub>2</sub> O <sub>4</sub> S m/z = 257.0596	- 1.2
177.1024 $\pm$ 0.0003	C <sub>10</sub> H <sub>13</sub> N <sub>2</sub> O m/z = 177.1028	- 0.4
92.0497	C <sub>6</sub> H <sub>6</sub> N m/z = 92.0500	- 0.3



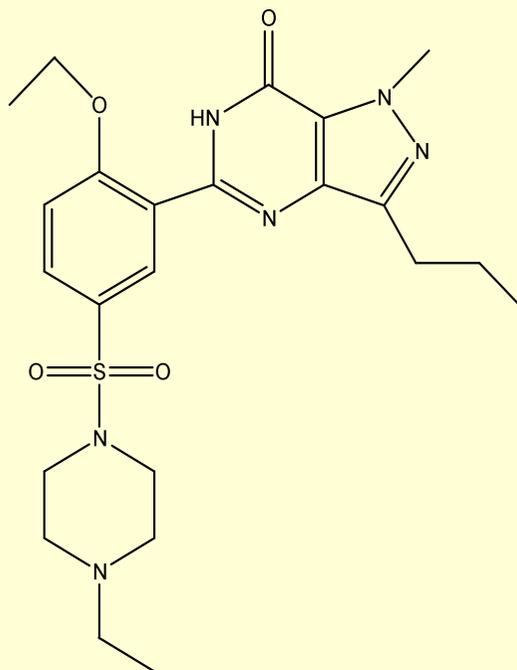
present in drainage channels on the Ebro delta

# Phosphodiesterase type V inhibitors

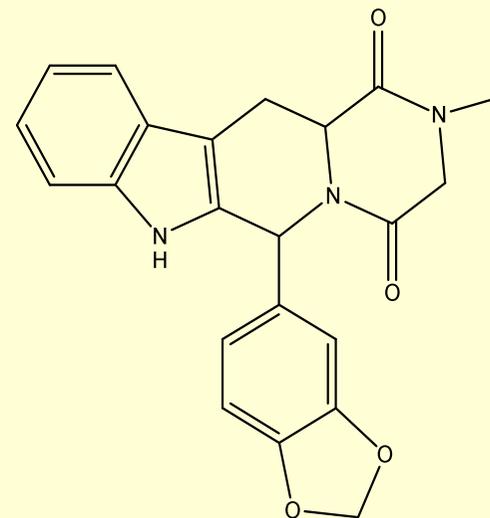
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Sildenafil

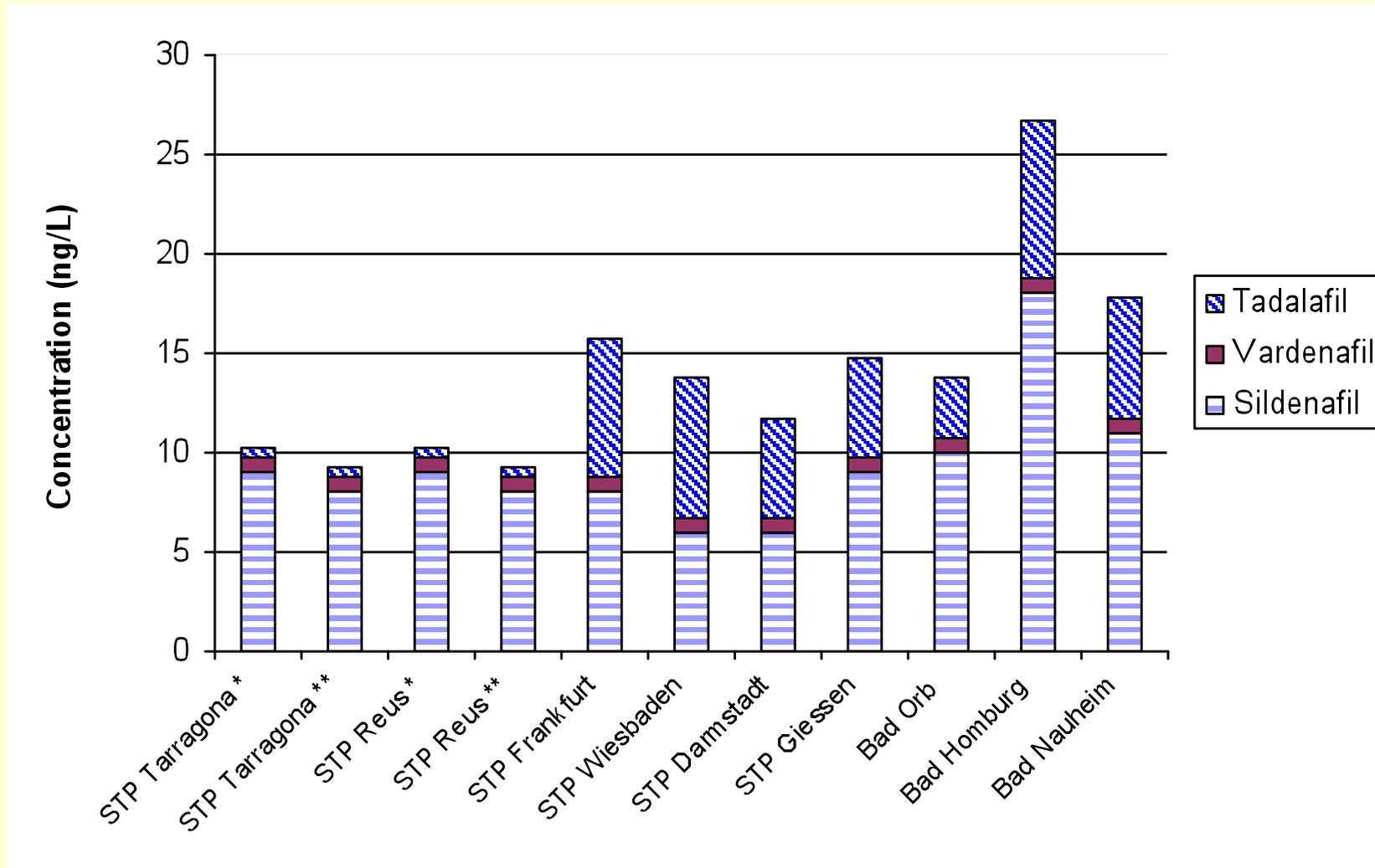


Vardenafil



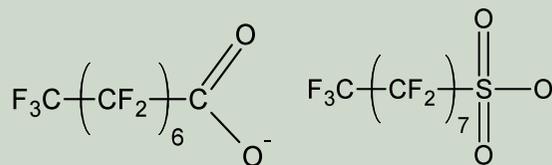
Tadalafil

# Phosphodiesterase type V inhibitors



# Fluorinated Surfactants

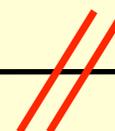
## Perfluorinated Surfactants



- Tremendous decrease in surface tension
- High stability towards oxidation, heat, ...
  - Special applications

- Extreme environmental persistence
  - Ubiquitous detection
  - (Eco)toxicological potential

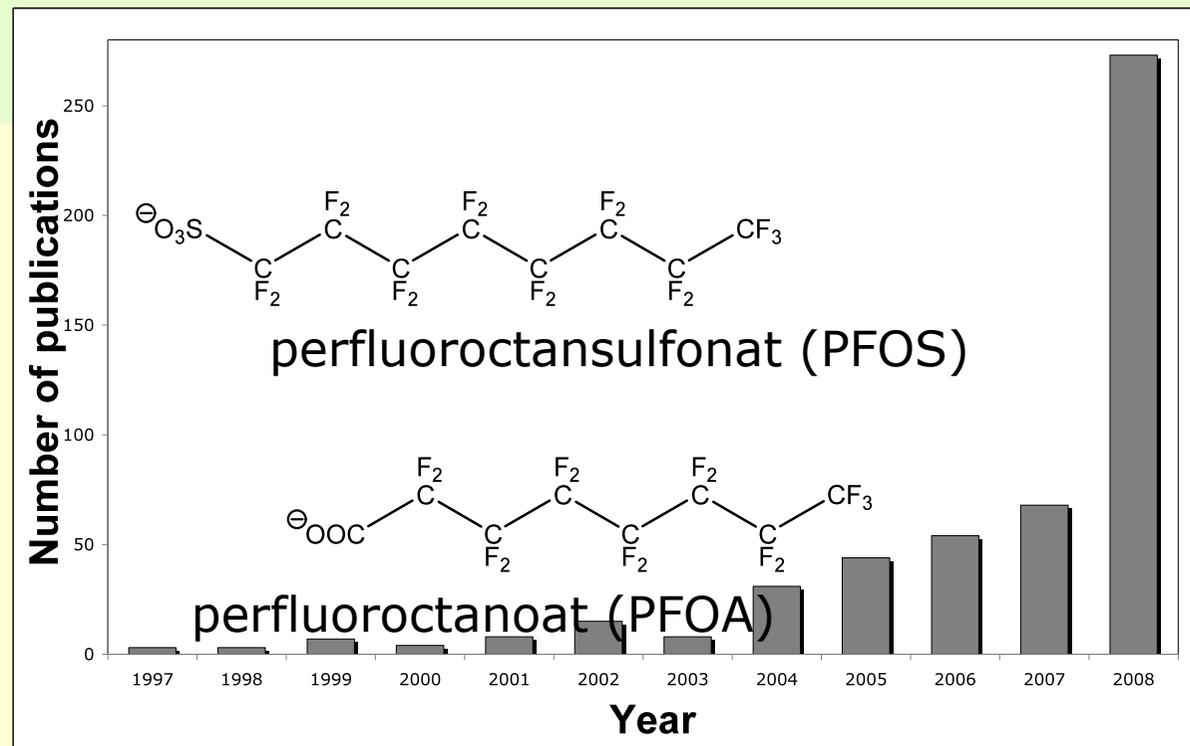
Adequate  
Substitutes ?



# Fluorinated surfactants

- surfactants
  - water soluble
  - resistant (hydrolysis/photolysis)
  - non biodegradable
  - non volatile
  - ubiquitous
  - bioaccumulative

PFOS included on the OSPAR list



➤ Can perfluorinated surfactants be replaced?

# Test System

## Fixed-Bed Bioreactor

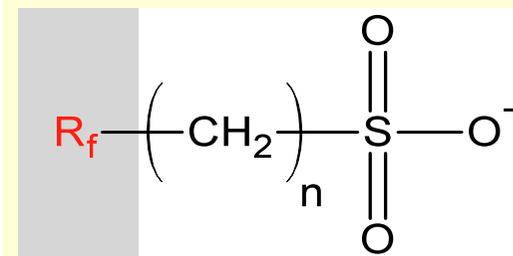
- Column used with porous glass spheres
  - ▶ Formation of a biofilm
- Filled with waste water effluent or river/creek water
  - ▶ Ubiquitous microorganisms
- Samples taken via three-way-valve



# Investigated compounds

$R_f$	n
$CF_3-O\text{~~~~}$	6, 7, 9, 10, 12, 14
$F_3C-C_6H_4-O\text{~~~~}$	9
$F_3C-N\text{~~~~}$ $F_3C$	7 - 13
$F_3C-N(CH_2)_3-O\text{~~~~}$ $F_3C$	6
$F_3C-CH_2-CH_2-O\text{~~~~}$	10
$F_3C-CH_2-CH_2-S\text{~~~~}$	11

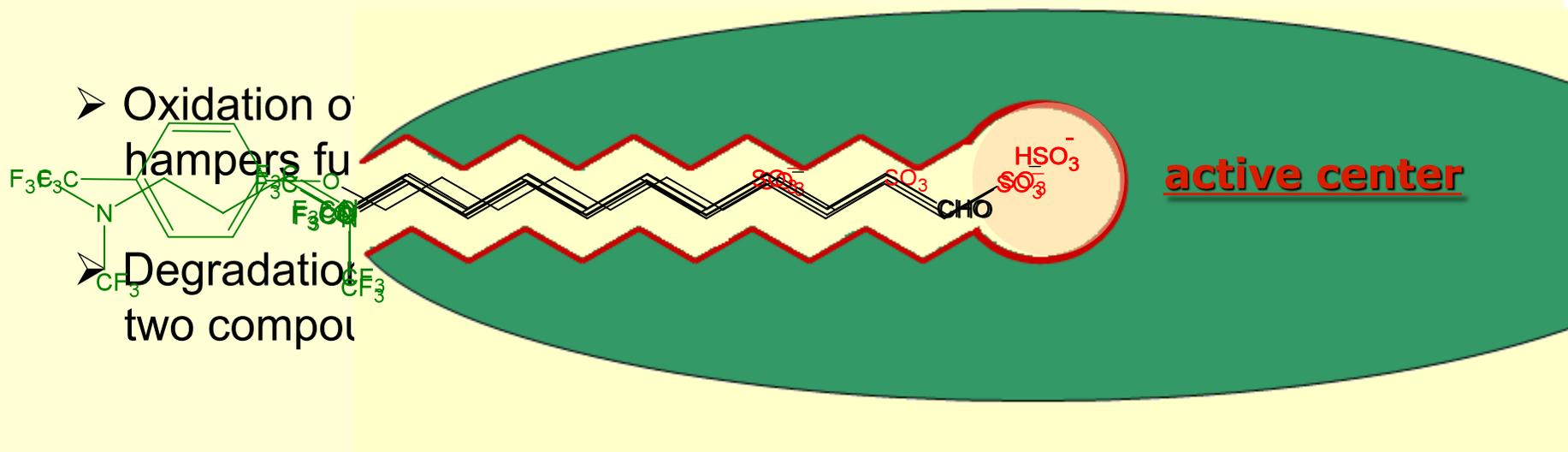
?  
⋮  
degradable



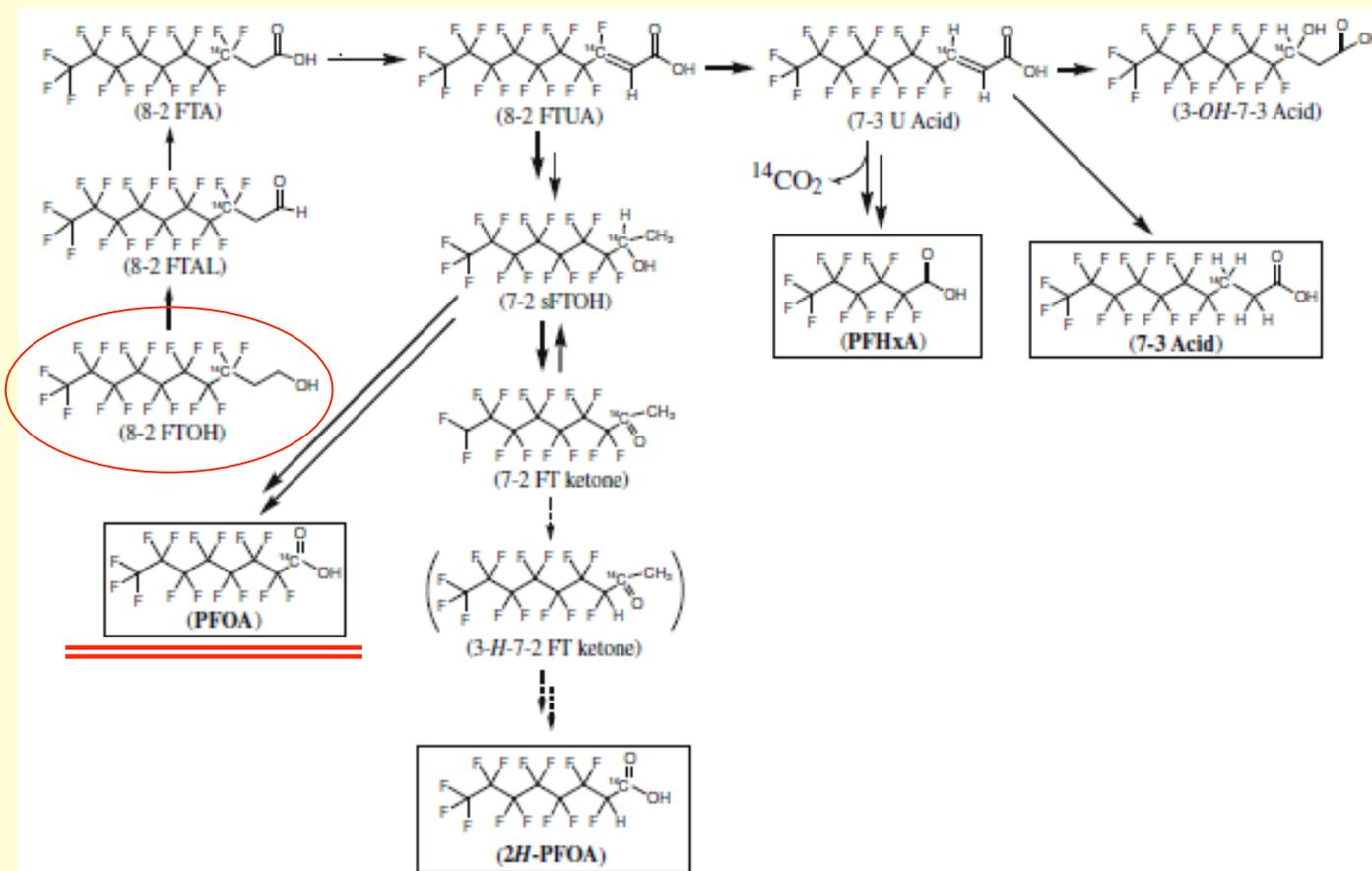
# Conclusions

- There are two different primary degradation steps: desulfonation and alkyl chain oxidation
- Desulfonation as a primary degradation step depends on the alkyl chain length

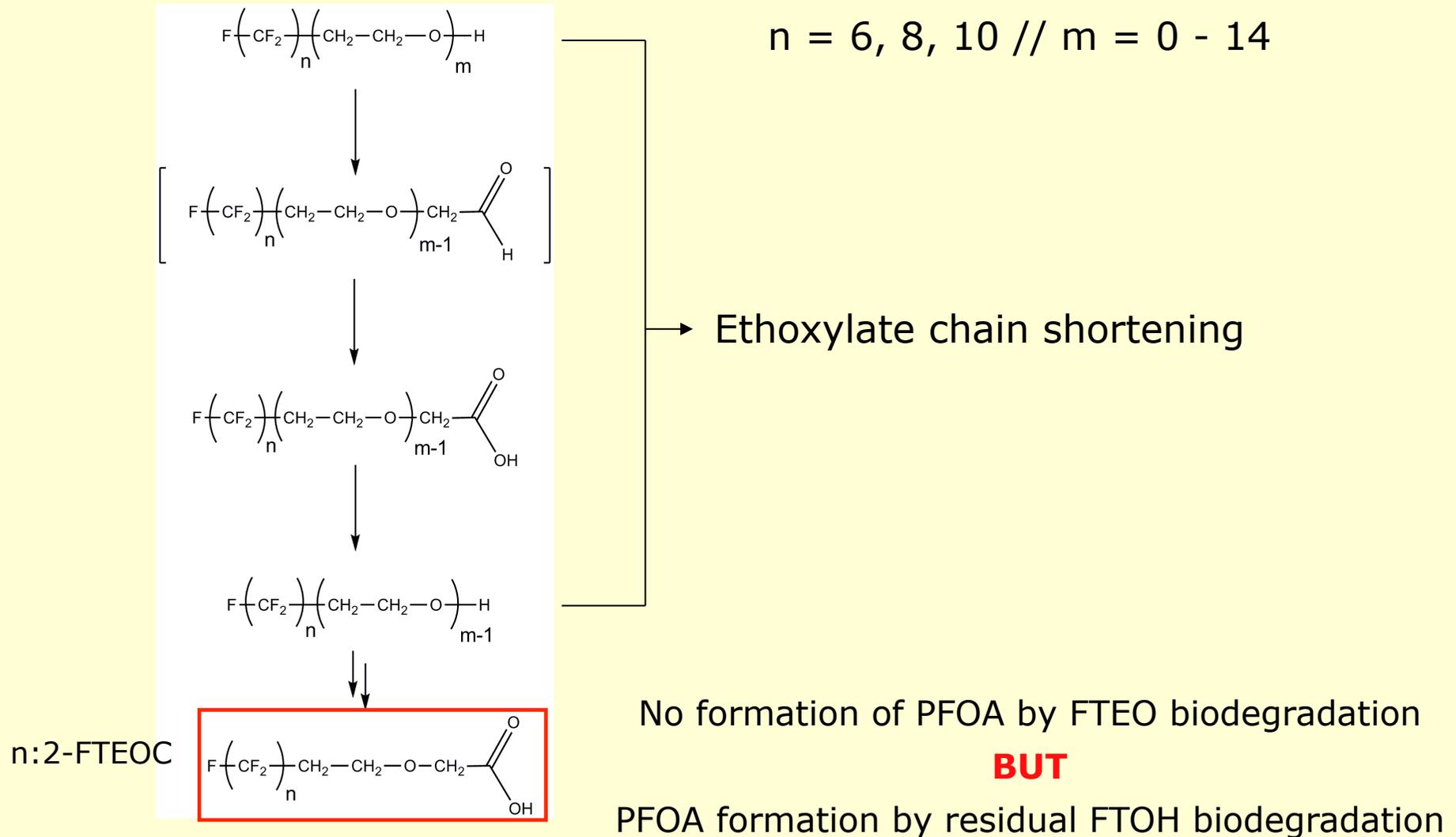
- Oxidation of hampers fu
- Degradation two compo



# Biodegradation of Fluorotelomer alcohols (8:2-FTOH)

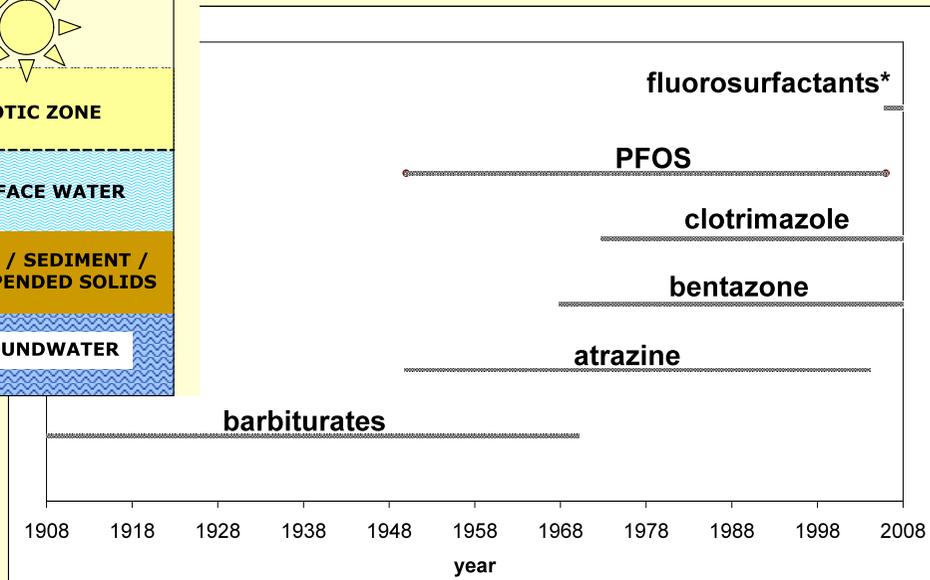
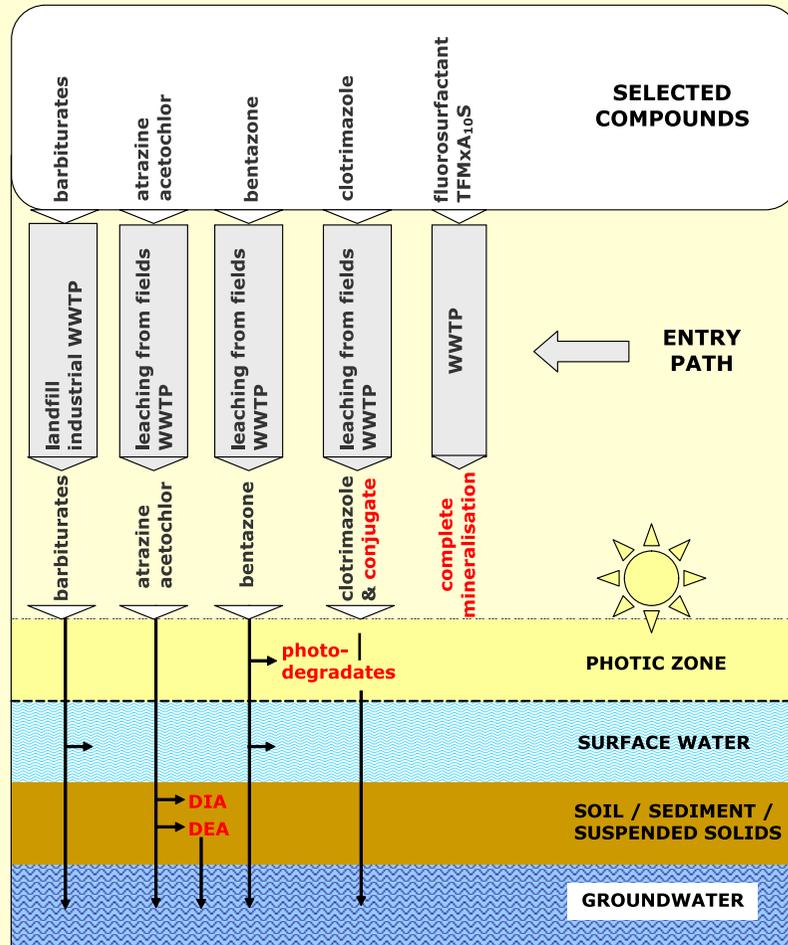


# Biodegradation pathway of fluorotelomer ethoxylates (FTEO)





# Final remarks



# Acknowledgement

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- AQUATERRA (Project number 505428 GOCE)
- Hessian Agency for Environment and Geology

**Thank you for your attention!**